Metanol CH₃OH

ALLAM 2019 Astrochemistry LLAMA Meeting

Observational Astrochemistry Astrochemistry in the Laboratory Astrochemistry with LLAMA • Theoretical Astrochemistry • Astrochemistry and the Origin of Life

Glicoaldeido

Metil-acetileno CH₃CCH

Observações com APEX J=10-9 170 GHz K-lader Traçador de gás frio (50 K) G331 AUGUST 8-9, 2019 IAG-USP, São Paulo - Brazil www.allam2019.iag.usp.br

burnersdore de São Pare-Instituiro de Quinica Alexandre de Quinica Alex

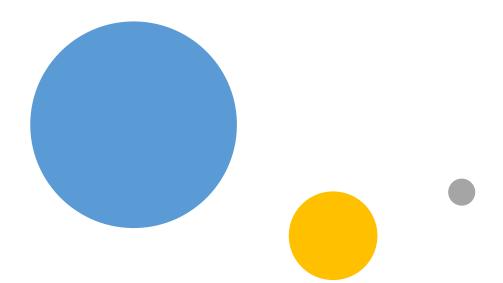
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 $H_{2}O$

CS

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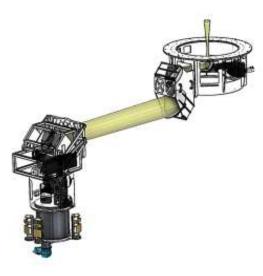
The situation of LLAMA in August 2019



ASTRO Electro-Mechanical Engineering LLC USA - designed the mechanical parts and set of mirrors Based on the optical design of Jacob Kooi













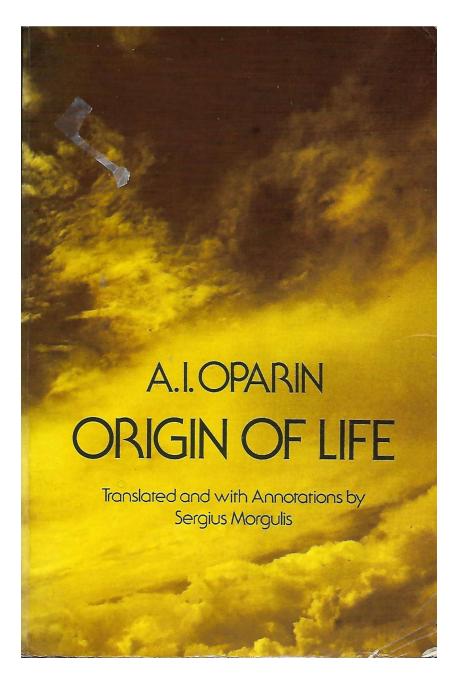


Situation of LLAMA in August 2019

A new organization is being set up in Argentina. LLAMA and QUBICS (An Italian-French-Argentinean project for Cosmology) will be under a single organization of the site, under the direction of Dr. Alberto Etchegoyen who was responsible on the Argentinean side for the installation of the Pierre Auger Cosmic Rays Observatory

- The plans for a power plant for the joint project, using gas from a gas duct that passes close to the site are detailed, and they are taking the first steps to start it
- The organization is hiring a site manager
- On the Brasilian side: The NACOS (Nasmith Cabin Optical System) -see photos -is nearly ready, now in the phase of alignment of mirrors inside it.
- A calibration Loads System with two temperature-controlled loads is being developed in collaboration with the University of Concepción (Rodrigo reeves). It will be ready in 6 months
- The two receivers for first light (Band 5 and Band 9), and the cryostat in which the will be installed are ready and waiting in the NOVA labs (Groningen)

- A holographic system for adjusting the panels of the antenna is being developed. The holographic receiver will be on loan form ALMA, and the transmitter is being developed in São Paulo
- An optical telescope for first pointing adjustments, with camera, external protection etc has been developed in Sao Paulo
- The last two items are needed for the acceptance tests of the antenna, as soon as it is mounted
- The big problem that I see: we cannot set up a clear schedule, because the Argentinean side has not yet guaranteed the funds for the construction of the concrete base on which the antenna will be mounted, and other investment needed on the site. Maybe finding new partners could be a solution.



The Origin of Life

By A.I Oparin, *Associate Director, Bio-Chemical Institute* U.S.S.R. Academy of Science

Copyright 1938 !!!!

But Oparin cites a refrence, **H.F. Osborn,** The origin and Evolution of life London, 1918 !!!

Oparin did not think about molecules in space.

He considered **natural mechanisms in the primitive Earth**. Like volcanism, reactions at high temperatures, Superheated steam

He mentions that it was demontrated that methane, heated at 1000° C, without catalyst changes to acetylene

 $2 \operatorname{CH}_4 \rightleftharpoons \operatorname{CH} \equiv \operatorname{CH} + 3 \operatorname{H}_2 - 91 \operatorname{Cal}.$ Methane Acetylene

However, in the presence of a sufficient amount of water vapor, we have not merely the formation of unsaturated hydrocarbons but also of the oxidized hydroxy derivatives.

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\mathrm{CH} \equiv \mathrm{CH} + \mathrm{H}_2\mathrm{O} \to \mathrm{CH}_3\mathrm{COH} \qquad \text{Acetaldehyde}
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In the presence of oxyde of Iron this reaction can take place at 300 C

In addition, OPARIN pays much attention to the reactions that take place inside living cells:

- 1) condensation (lenthening of carbon chain)
- 2) splitting of chains (reverse of previous one)
- 3)polimerization (union between organic molecules through atom of oxygen or nitrogen)
- 4) hydrolysis (reverse process of above splitting unions)
- 5) oxydation-reduction mechanisms

In this reaction two molecules of acetaldehyde become welded together, so to speak, to form a single molecule of aldol, a compound consisting of a four carbon atom chain. as was shown previously, glycol aldehyde is produced which, by an oxidation-reduction reaction, yields glycol and glycolic acid. The latter reacting with ammonia gives glycine, the simplest of the amino acids:

- 1. $2 \operatorname{CH}_2 \operatorname{OH} \cdot \operatorname{COH}_{+} \operatorname{H}_2 \operatorname{OH} \cdot \operatorname{CH}_2 \operatorname{OH} \cdot \operatorname{CH}_2 \operatorname{OH} + \operatorname{CH}_2 \operatorname{OH} \cdot \operatorname{COOH}_{\operatorname{Glycol}} \operatorname{Glycolic} \operatorname{acid}_{\operatorname{Glycolic}}$
- 2. $NH_3 + HO \cdot CH_2 \cdot COOH = NH_2 \cdot CH_2 \cdot COOH + H_2O$ Ammonia Glycolic acid Glycine

Other amino acids are formed in a similar manner