Aline Ramos Ribeiro, Diana P. P. Andrade, Heloisa M. Boechat-Roberty Observatório do Valongo, Universidade Federal do Rio de Janeiro



ALLAM: Astrochemistry LLAMA Meeting IAG/USP, August 9, 2019



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INTRO

METHODOLOGY RESULTS CONCLUSIONS

FUTURE WORK

### WHAT'S INTERESTING ABOUT TITAN

Shares many similatiries with Earth

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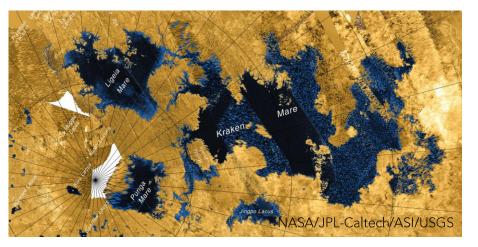
INTRO

METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

# WHAT'S INTERESTING ABOUT TITAN

Shares many similatiries with Earth

Lakes



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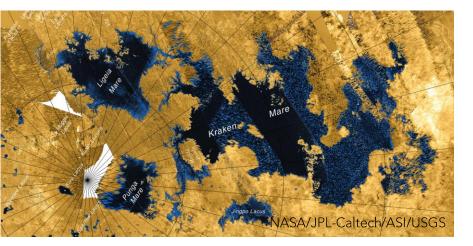
INTRO

METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

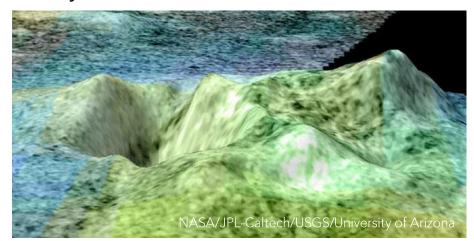
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Cryovolcanism



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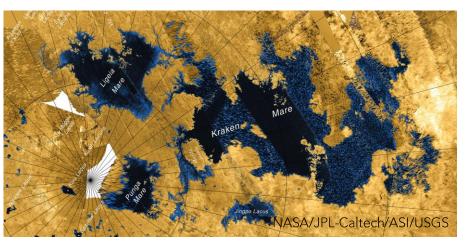
INTRO

METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

# WHAT'S INTERESTING ABOUT TITAN

Shares many similatiries with Earth

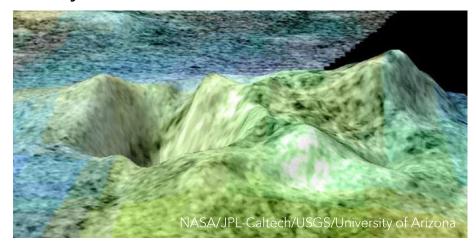
Lakes



Dense atmosphere



Cryovolcanism



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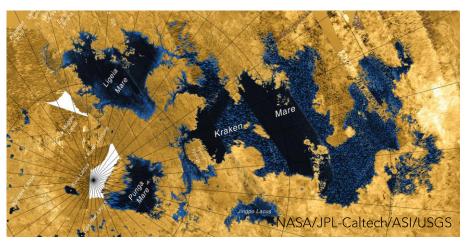
INTRO

METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

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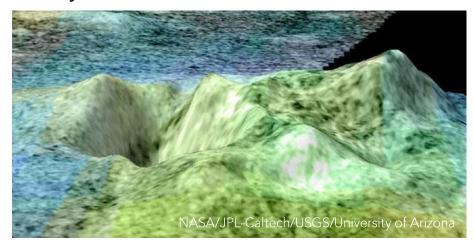
Lakes



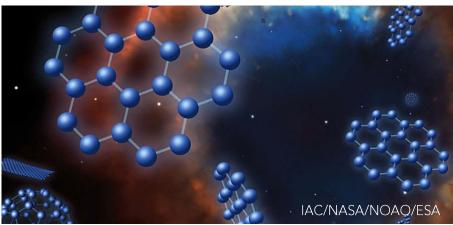
Dense atmosphere



Cryovolcanism



• Complex organic molecules (COMs)

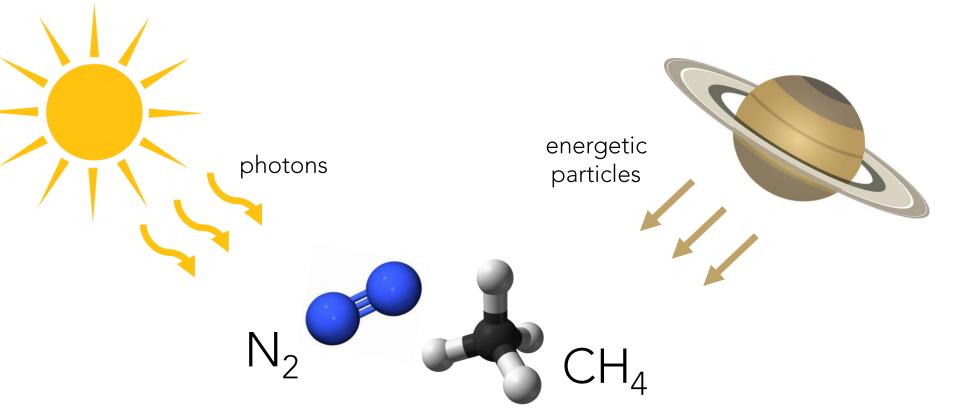


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INTRO

METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

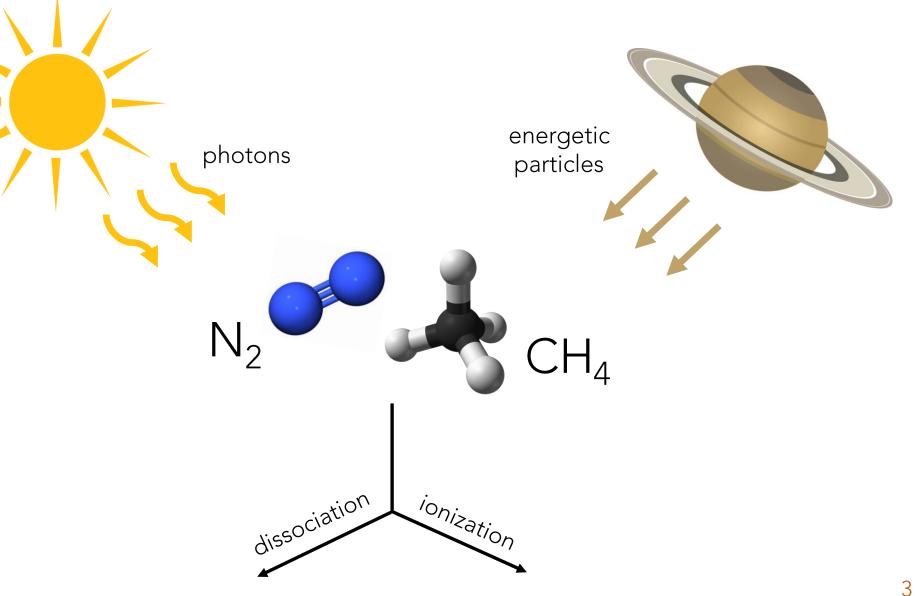
#### CHEMISTRY IN TITAN



INTRO

METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

#### CHEMISTRY IN TITAN

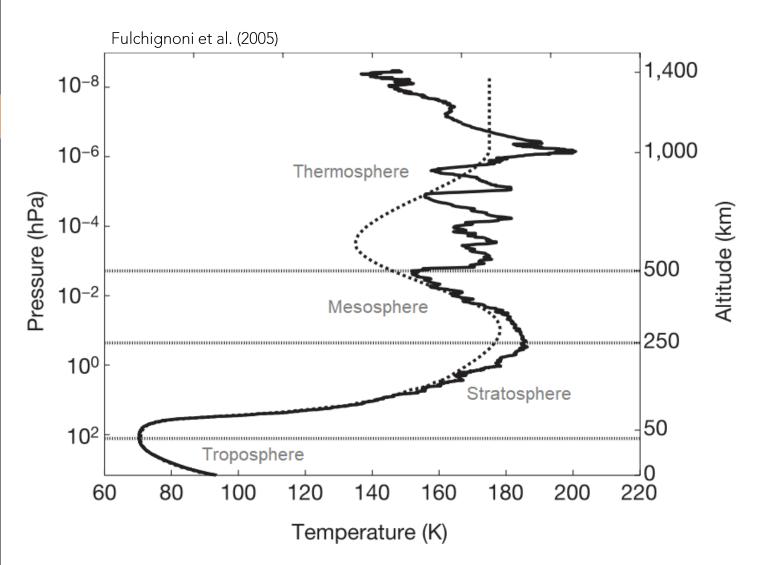


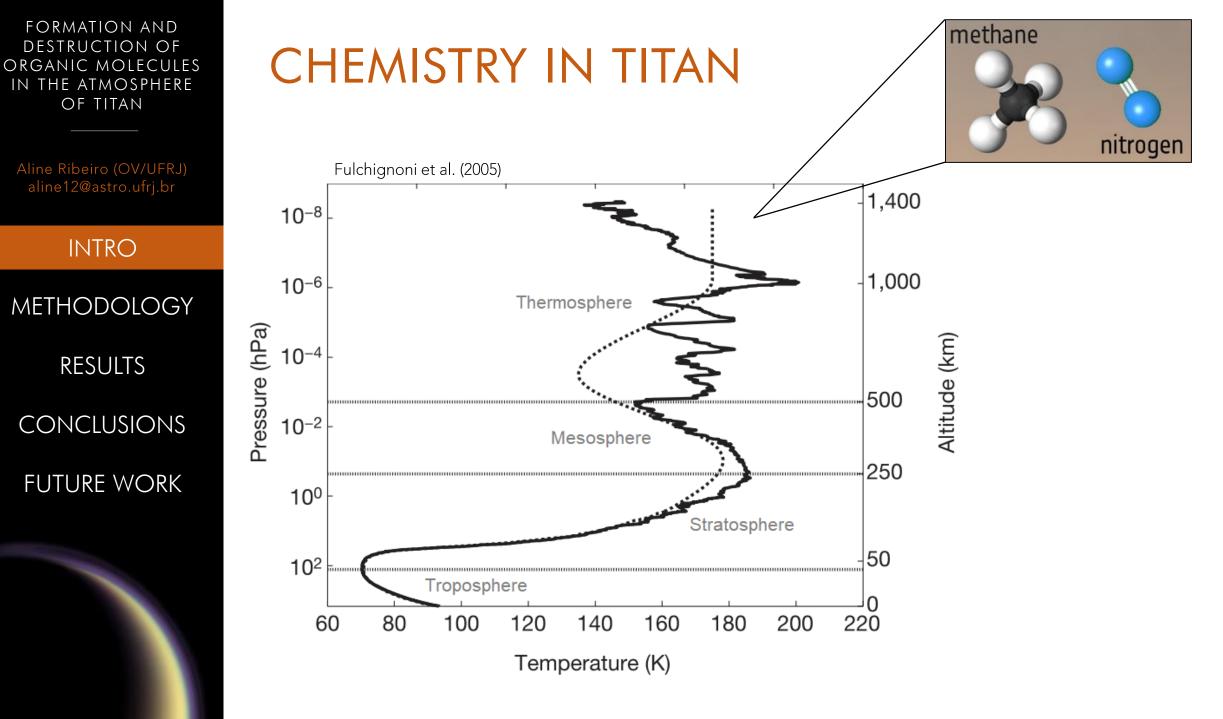
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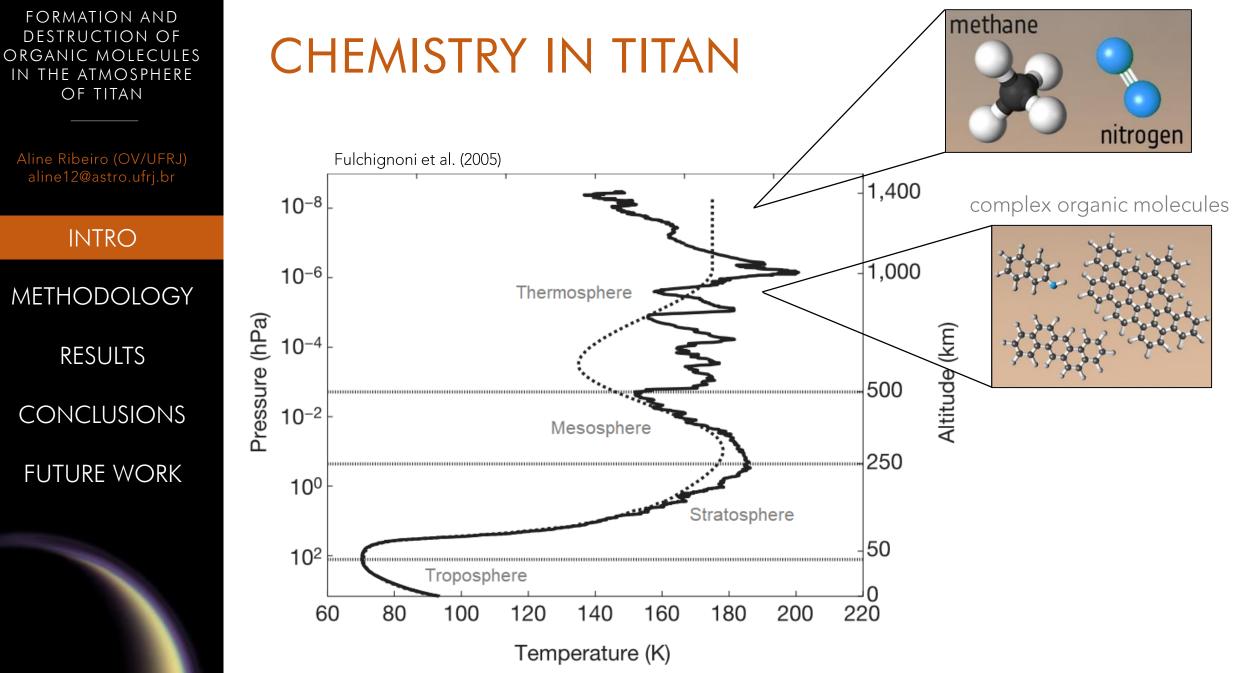
INTRO

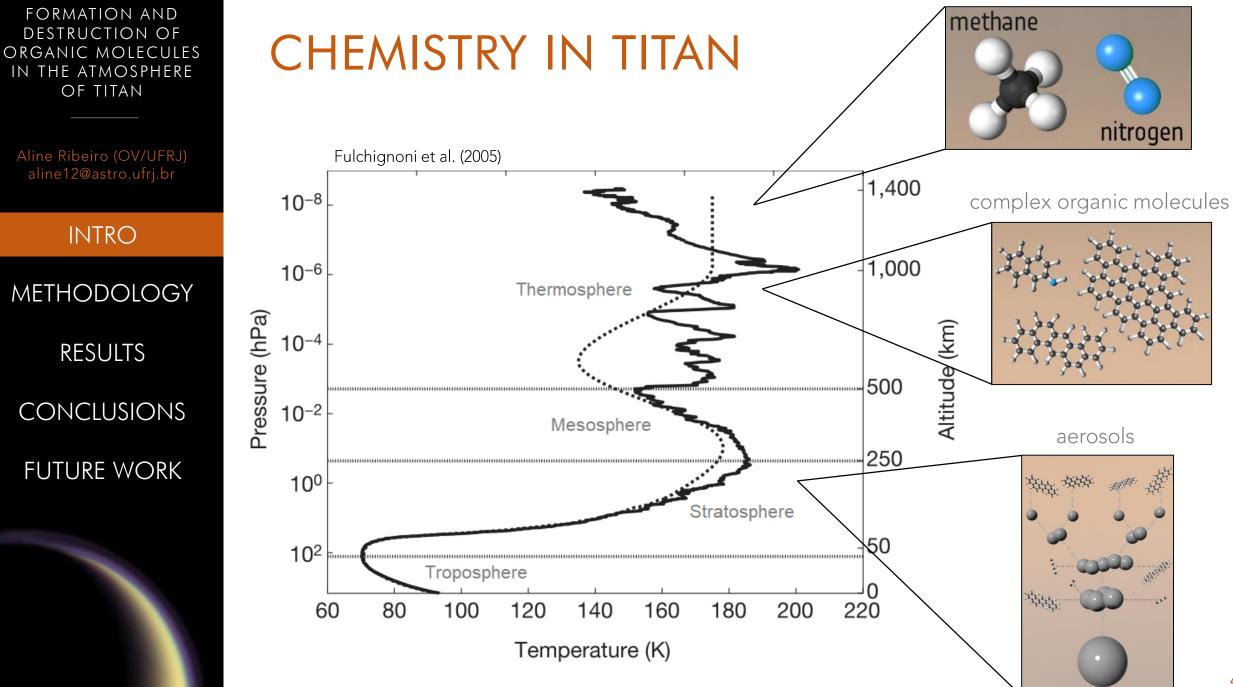
METHODOLOGY RESULTS CONCLUSIONS FUTURE WORK

# CHEMISTRY IN TITAN









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INTRO

METHODOLOGY

RESULTS CONCLUSIONS FUTURE WORK

# PHOTOCHEMICAL MODEL

- 1D and stationary
- Simulates the formation and destruction of molecules in Titan's upper atmosphere

AstroReactions and ReactionEquations by Pinotti & Boechat-Roberty (2016)

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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

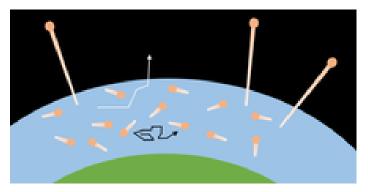
FUTURE WORK

# PHOTOCHEMICAL MODEL

- 1D and stationary
- Simulates the formation and destruction of molecules in Titan's upper atmosphere

AstroReactions and ReactionEquations by Pinotti & Boechat-Roberty (2016)

Upper atmosphere only  $\longrightarrow$  above the exobase (1450 km)



atmospheric escape

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INTRO

METHODOLOGY

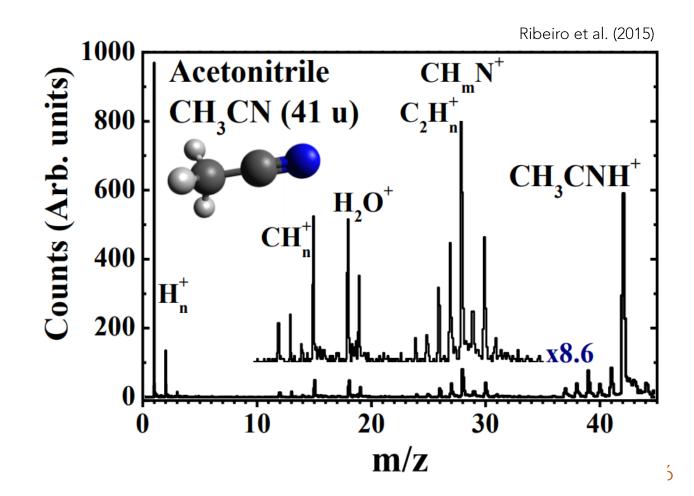
RESULTS CONCLUSIONS

FUTURE WORK

#### PHOTOCHEMICAL MODEL

selection of species

- literature
- group works (e.g. Ribeiro et al. 2015)



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INTRO

#### METHODOLOGY

results conclusions

FUTURE WORK

### PHOTOCHEMICAL MODEL

selection of species

- literature
- group works (e.g. Ribeiro et al. 2015)
- UMIST database (McElroy et al. 2013)

	ST RATE12 ochemistry.net
Home Downloads Species Search	
common H <sub>2</sub> CO H	UMIST RATE2012 / astrochemistry.net
он нсо⁺	Welcome to the 2012 edition of The UMIST Database for Astrochemistry.
С Н <sub>3</sub> +	This is the 5th public release of the database.
C <sup>+</sup> H₂O	The database download files and the paper are available from the download section.
e	Recent updates
in RATE12 C C C C C C C C C C C C C C C C C C C	21/03/16: Python scripts by Paul Woods that take output from UDfA chemical models and generate input files for popular radiative transfer codes. Available in the <u>download</u> section.

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INTRO

METHODOLOGY

results conclusions

FUTURE WORK

### PHOTOCHEMICAL MODEL

selection of species

• literature

- group works (e.g. Ribeiro et al. 2015)
- UMIST database (McElroy et al. 2013)
- C,H,O,N molecules only

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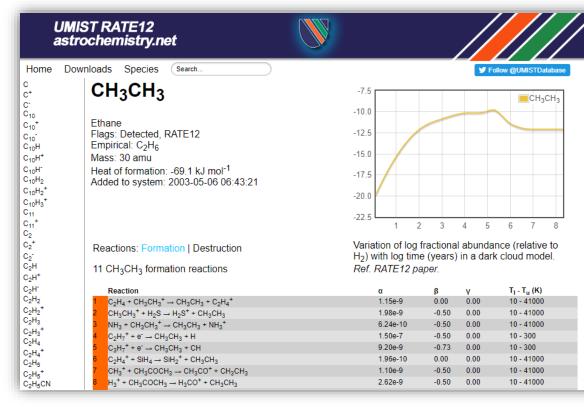
INTRO

#### METHODOLOGY

RESULTS CONCLUSIONS FUTURE WORK

#### PHOTOCHEMICAL MODEL





 UMIST database (McElroy et al. 2013)

#### 7

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INTRO

METHODOLOGY

RESULTS CONCLUSIONS FUTURE WORK

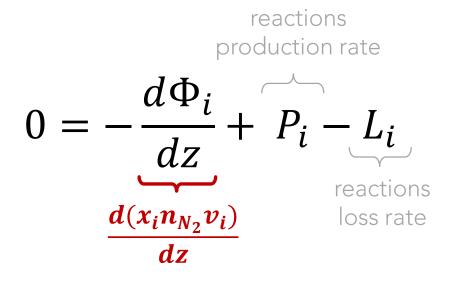
# PHOTOCHEMICAL MODEL

reactions list for each specie

solving continuity equations

Continuity equation for a chemical specie i:

species



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INTRO

METHODOLOGY

RESULTS CONCLUSIONS FUTURE WORK

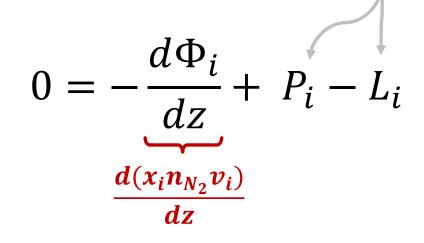
# PHOTOCHEMICAL MODEL

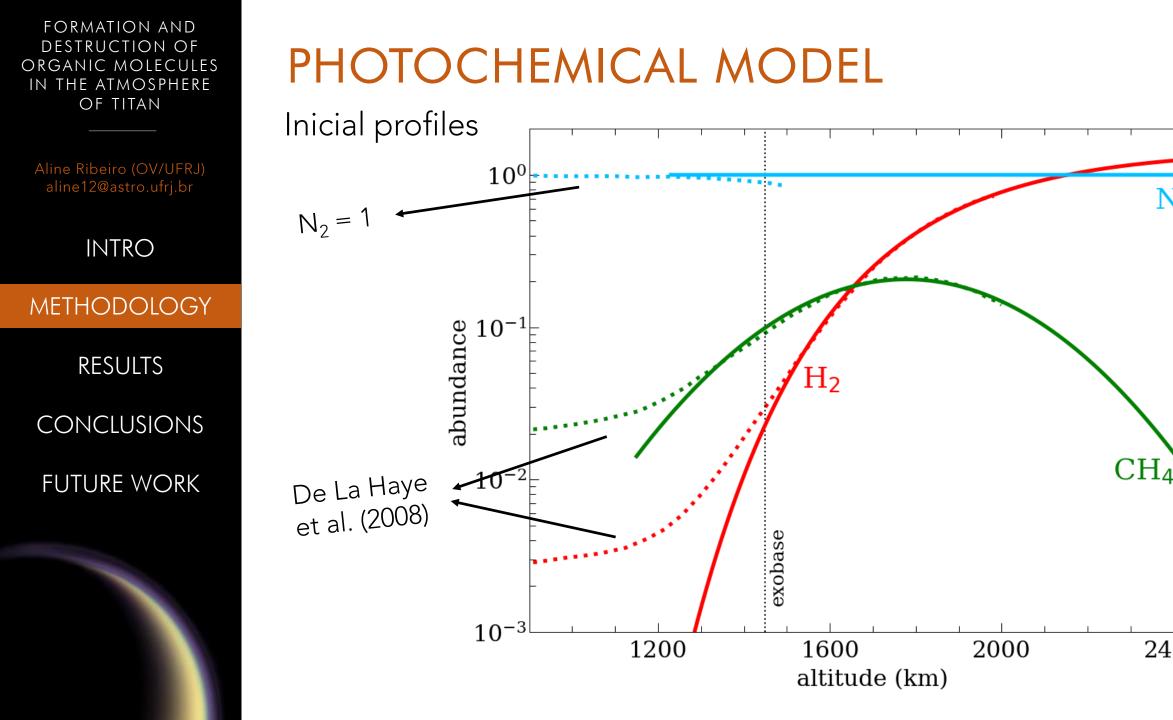
reactions list for each specie solving continuity equations

Continuity equation for a chemical specie i:

species

P (solar flux)
p (magnetospheric flux)





 $N_2$ 

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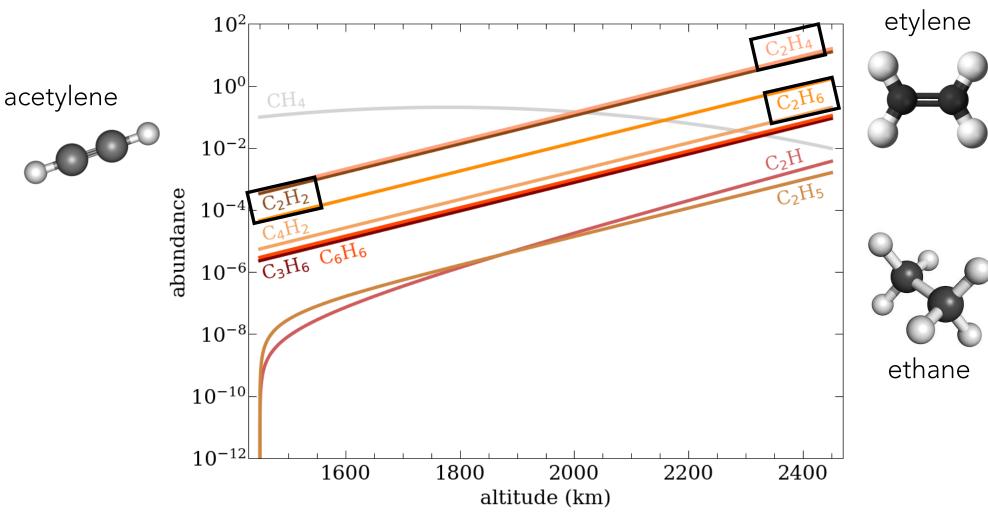
INTRO

METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

#### SIMULATING MOLECULE ABUNDANCES Hydrocarbons



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INTRO

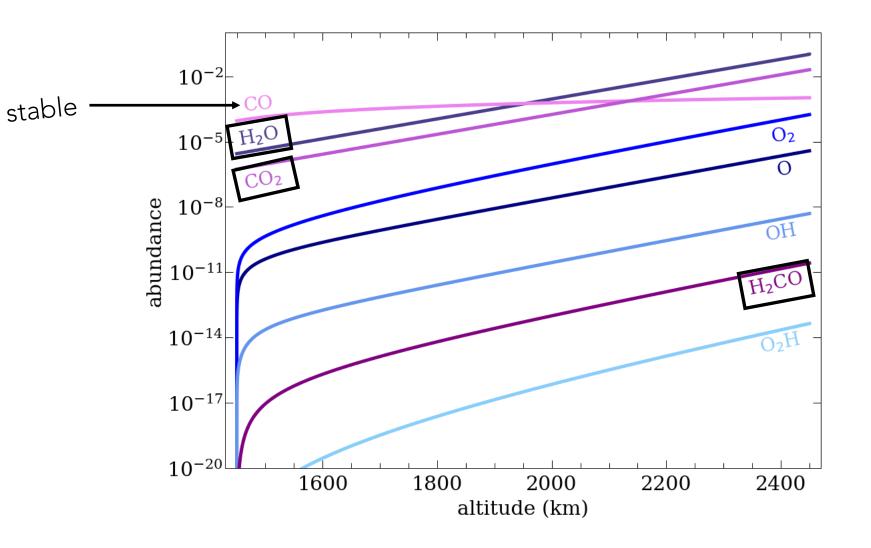
METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

# SIMULATING MOLECULE ABUNDANCES

Oxygen-bearing molecules



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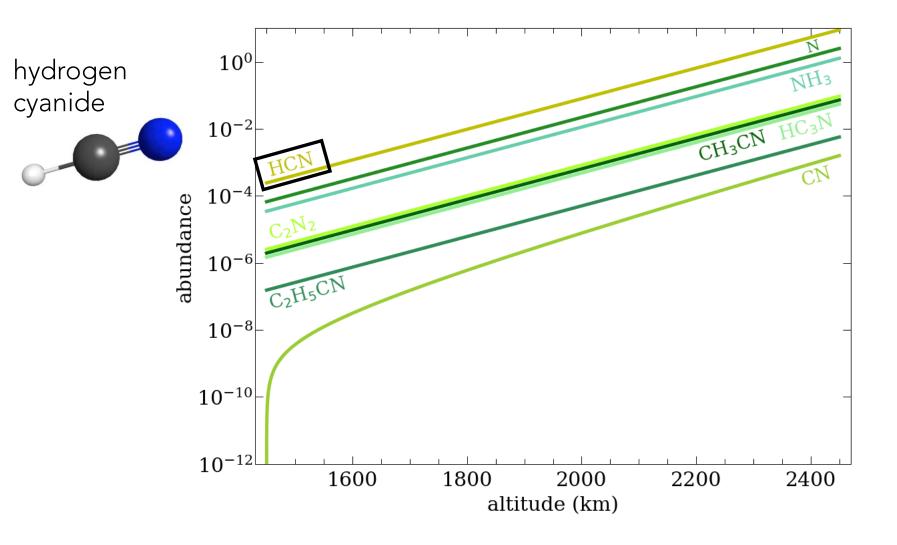
METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

# SIMULATING MOLECULE ABUNDANCES

Nitrogen-bearing molecules



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INTRO

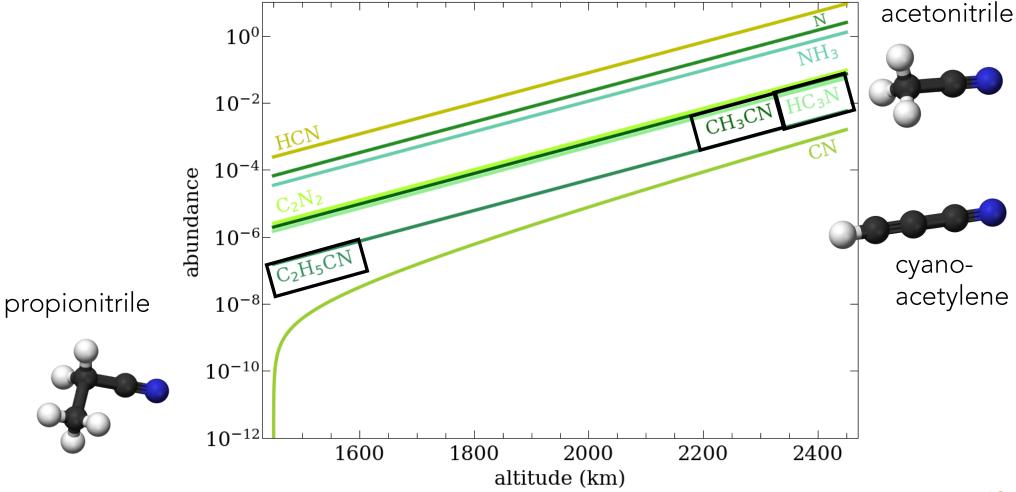
METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

# SIMULATING MOLECULE ABUNDANCES

Nitrogen-bearing molecules



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INTRO

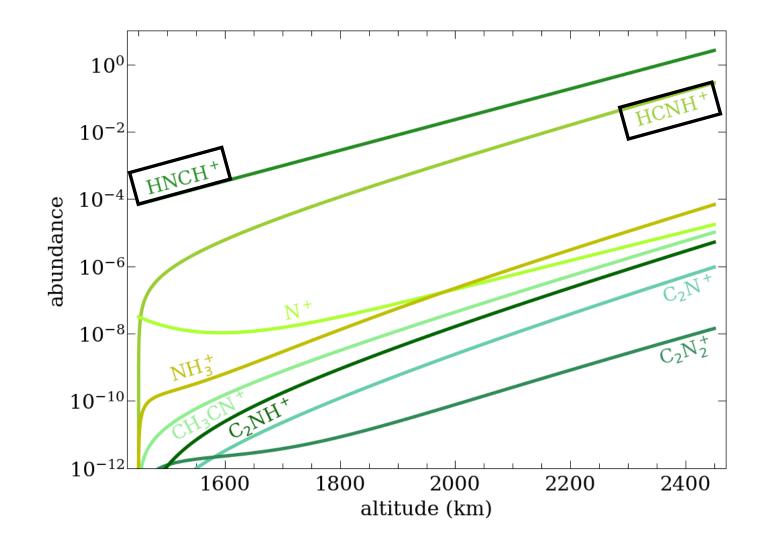
METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

# SIMULATING MOLECULE ABUNDANCES

Nitrogen-bearing molecules



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INTRO

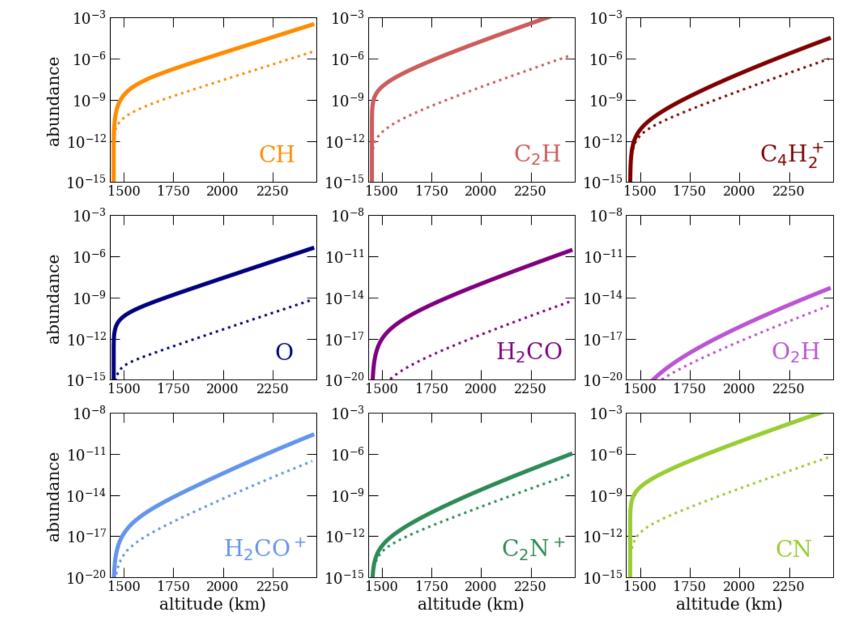
METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

# MAGNETOSPHERIC INFLUENCE

magnetosphere no magnetosphere



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INTRO

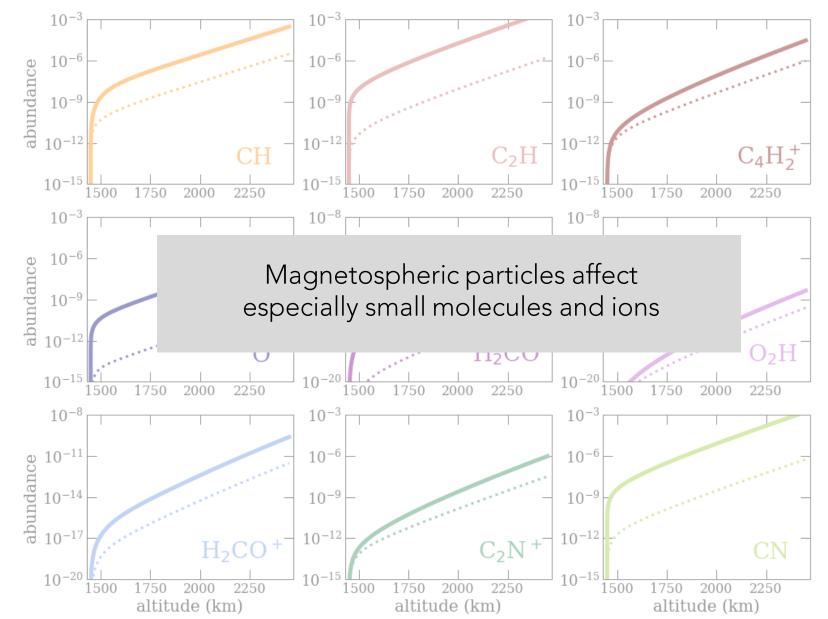
METHODOLOGY

RESULTS

CONCLUSIONS FUTURE WORK

### MAGNETOSPHERIC INFLUENCE

magnetosphere
 no magnetosphere



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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

FUTURE WORK

# CONCLUSIONS

Molecule abundances in agreement with literature considering **atmospheric escape** processes

3 most abundant hydrocarbons: C<sub>2</sub>H<sub>2</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>6</sub>
 o not influenced by O-bearing and N-bearing species

Evidence of nitrileso precursors of prebiotic molecules

Evidence of magnetospheric influence in the formation of compounds

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INTRO

METHODOLOGY

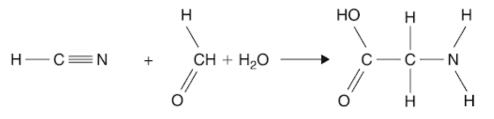
RESULTS

CONCLUSIONS

FUTURE WORK

#### DISCUSSIONS

Formation of glycine: C<sub>2</sub>H<sub>5</sub>NO<sub>2</sub>
 from CH<sub>2</sub>OH or CH<sub>3</sub>COOH (e.g. Pilling et al. 2011)
 from HCN + HCOH + H<sub>2</sub>O (e.g. Wayne, 2018)



Hydrogen cyanide Formaldehyde Water

Glycine

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INTRO

METHODOLOGY

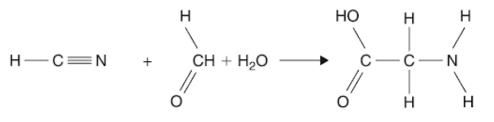
RESULTS

CONCLUSIONS

FUTURE WORK

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Formation of glycine: C<sub>2</sub>H<sub>5</sub>NO<sub>2</sub>
 from CH<sub>2</sub>OH or CH<sub>3</sub>COOH (e.g. Pilling et al. 2011)
 from HCN + HCOH + H<sub>2</sub>O (e.g. Wayne, 2018)



Hydrogen cyanide Formaldehyde Water

Glycine

Formation of adenine: C<sub>5</sub>H<sub>5</sub>N<sub>5</sub> • experimental (e.g. Pilling et al. 2009) • from HCN 5 H-c=N ]

HYDROGEN CYANIDE

ADENINE

 $NH_2$ 

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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

FUTURE WORK

# PERSPECTIVES

- Upgrade the model
  - Titan's induced magnetic field?
  - o complete atmosphere description
  - Inclusion of new compounds
     e.g. PAHs from C<sub>6</sub>H<sub>6</sub>
    - S and P molecules

New database (e.g. KIDA)

- Laboratory experiments
  - verify the stability of COMs, especially the ones with astrobiological interest

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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

FUTURE WORK

#### MORE INFO:





@exoplanetaline



astrotubers

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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

FUTURE WORK

#### **APPENDICE:**

Constantes de reação:

entre dois corpos: 
$$k = \alpha \left(\frac{T}{300}\right)^{\beta} exp\left(-\frac{\gamma}{T}\right)$$

• com prótons de raios cósmicos:  $k_{CRP} = \alpha$ 

- com fótons de raios cósmicos:  $k_{CRPHOT} = \alpha \left(\frac{T}{300}\right)^{\beta} \frac{\gamma}{1-\omega}$ 
  - com fótons do meio interestelar:  $k_{PHOTON} = \alpha exp(-\gamma A_V)$

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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

FUTURE WORK

#### **APPENDICE:**

Fator de correção para o fluxo de radiação solar:

• entre dois corpos: 
$$k = \alpha \left(\frac{T}{300}\right)^{\beta} exp\left(-\frac{\gamma}{T}\right)$$

• com prótons de raios cósmicos:  $k_{CRP} = \alpha$ 

• com fótons de raios cósmicos:  $k_{CRPHOT} = \alpha \left(\frac{T}{300}\right)^{\beta} \frac{\gamma}{1-\omega}$ 

com fótons do meio interestelar:  $k_{PHOTON} = \alpha \exp(-\gamma A_V) P$ 

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INTRO

METHODOLOGY

RESULTS

CONCLUSIONS

FUTURE WORK

#### **APPENDICE:**

Fator de correção para o fluxo de plasma da magnetosfera de Saturno:

• entre dois corpos: 
$$k = \alpha \left(\frac{T}{300}\right)^{\beta} exp\left(-\frac{\gamma}{T}\right)$$

• com prótons de raios cósmicos:  $k_{CRP} = \alpha p$ 

- com fótons de raios cósmicos:  $k_{CRPHOT} = \alpha \left(\frac{T}{300}\right)^{\beta} \frac{\gamma}{1-\omega} p$ 
  - com fótons do meio interestelar:  $k_{PHOTON} = \alpha \exp(-\gamma A_V) P$